## **Supporting Information**

## Formation of a Silicon Layer by Electroreduction of SiO<sub>2</sub> Nanoparticles in

## CaCl<sub>2</sub> Molten Salt

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**Figure S1.** A cyclic voltammogram on a Mo electrode (1 mm dia. wire,  $0.22 \text{ cm}^2$ ) at a scan rate of 100 mV/s with 10 mA/cm<sup>2</sup> of current density passing on dynamic reference electrode (1 mm dia. Mo wire,  $0.22 \text{ cm}^2$ ) in 850 °C CaCl<sub>2</sub> melt.

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**Figure S2.** Cyclic voltammograms on a Mo electrode (a) contacted with (0.3 cm wide and 0.5 cm long) quartz piece and (b) with 0.2 M Na<sub>2</sub>SiO<sub>3</sub> in 850 °C CaCl<sub>2</sub> melt.



**Figure S3.** A SEM image of Si deposited on Mo grown in 850 °C CaCl<sub>2</sub> melt containing 0.2 M  $Na_2SiO_3$  for 500 s. EDS analyses show that the Si film on Mo is composed of Si 48 at%, O 8 at%, C 19 at%, Na < 1 at%, and Mo 24 at%.



Figure S4. Cyclic voltammograms on a Mo foil working electrode (0.3 cm width, 0.42 cm<sup>2</sup>) at a scan rate of 20 mV/s with SiCl<sub>4</sub> gas mixed with Ar carrier gas which is transferred directly to 850 °C CaCl<sub>2</sub> melt.