Supporting Information

Surface Interrogation Scanning Electrochemical Microscopy (SI-SECM) of Photoelectrochemistry at a W/Mo-BiVO₄ Semiconductor Electrode – Quantification of Hydroxyl Radicals During Water Oxidation

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Figure S1. Optical microscopic images of Au UME from side (a) and top (b), and the image of the W/Mo-BiVO₄ photoanode covered by the insulating layer (c). Optical surface profiler images of the photoanode (d) that is identical W/Mo-BiVO₄ shown in (c). The optical profiler was used to measure the exposed area of the photoanode (0.039 mm²) and to measure the thickness of the insulating Teflon layer on the photoanode (~ 20 μ m).



Figure S2. Cyclic voltammogram (CV) (a), approach curve (b, blue solid line) of Au UME, and SECM image of TC-SG experiments on W/Mo-BiVO₄ (c) in a 1 mM ferrocenemethanol (FcMeOH) and 0.1 M KCl aqueous solution. CV was measured in bulk solution with a scan rate of 20 mV s⁻¹ in (a). For the approach curve in (b), the tip moved toward the insulating substrate with the increment distance of 1 μ m per 0.2 s. The tip potential was 0.7 V (vs NHE). The theoretical curve (yellow dots) in (b) was obtained with R_g =2. (Sun, P.; Laforge, F. O.; Mirkin, M. V. *Phys. Chem. Chem. Phys.* **2007**, *9*, 802–823.) *d* is the distance between the tip and the substrate, and *r* is the radius of the tip, i.e. 25 μ m. For TC/SG-SECM in (c), tip potential was held at 0.3 V and the substrate was at 0.4 V under UV-Visible irradiation. Scan rate was 20 μ m

s⁻¹ with the increment distance of 4 μ m and the increment time was 0.2 s. The tip was placed about 12 μ m above the substrate.



Figure S3. Tip collection for oxygen reduction at Au UME (a) and substrate generation for water oxidation at W/Mo-BiVO₄ (b) in 0.1 M Na₂SO₄ aqueous solution (0.2 M phosphate buffer, pH 7). The potential of tip was at 0.5 V (vs NHE) and W/Mo-BiVO₄ was held at 0.6 V. The irradiation was switched on from 10 to 30 s with an ELH lamp (300 W) through the light guide (diameter of 3 mm) in the experimental configurations shown in Figure 1. Au UME tip with a diameter of 50 μ m was used and the area of W/Mo-doped BiVO₄ was larger than that used in Figure 8 to increase the generation current.



Figure S4. Chronoamperograms (CAs) of surface interrogation using Au UME on the W/Mo-BiVO₄ electrode with different decay times from 0.5 s to 15 s after UV-Visible irradiation for 5 s (yellow). CA of that without the irradiation is shown as a grey solid line. The potential of Au

UME was 0.5 V (vs NHE) and the W/Mo-BiVO₄ was held at 0.6 V during the irradiation. The potential and experimental configurations for the measurements was set as shown in Figure 1. Measurements were done in 1 mM K_2 IrCl₆ and 0.1 M Na₂SO₄ aqueous solution. Au UME was placed about 12 μ m above W/Mo-BiVO₄.



Figure S5. Interrogated charges of OH• at W/Mo-BiVO₄ with various irradiation durations of (a) 1, (b) 2, (c) 3, (d) 4, and (e) 5 s. For each measurement, four different decay times were allowed

from 0.5 to 4 s. Experimental configurations were identical with that shown in Figure 1 and Figure 6 except the duration of UV-Visible irradiation. The interrogated charge of the OH• was summarized in (f) for the different irradiation time. Electrode area of W/Mo-BiVO₄ or the collection factor of the interrogation experiments were not considered in the results shown in (f).



Figure S6. Linear sweep voltammogram of W/Mo-BiVO₄ with chopped light under UV-Visible irradiation in 0.1 M Na₂SO₄ and 0.1 M Na₂SO₄ aqueous solution (pH 7). Beam intensity was gradually increased from 100 (blue), 300 (green) to 400 mW cm⁻² (yellow) with full output from a xenon lamp. Scan rate was 20 mV s⁻¹.